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nonmetal atoms are soluble in water,

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entirely of nonmetal atoms are soluble in water, while others are not. Use the bonding models to explain why.

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Use the handout to answer these questions. 2. What do the pictures of the four models of bonding attempt to show? 3. Give the type of bonding for

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each substance described here. a. A substance made up entirely of metal atoms b. A substance made up of both metal and nonmetal atoms c. A substance made up entirely of nonmetal atoms LESSON 26 ACTIVITY

## **LESSON Electron Glue**

Use the handout to answer these



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questions. 2. What do the pictures of the four models of bonding attempt to show? 3. Give the type of bonding for each substance described here. a. A substance made up entirely of metal atoms b. A substance made up of both metal and nonmetal atoms c. A substance made up entirely of nonmetal atoms LESSON 26 AcTivITy

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## **LESSON 26 Electron Glue Date Period Bonding**

the bone represents the electron that is up for grabs, then when the big dog gains an electron he becomes negatively charged and the little dog who lost the electron becomes positively charged.  
"The two ions (that's where the name

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ionic comes from) are attracted very strongly to each other as a result of their opposite charges."

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to new positions the electron glue flows right along with the cations and holds the structure ...

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Model. Explain why the number ...  
Sample Answers All atoms in the same group have the same number of electrons in their outermost shell. As you move across a period, the number of electrons increases ... Electron Glue Bonding Purpose Period

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## **answers ...**

$n$   $l$   $m_l$   $m_s$ . State the four quantum numbers and the possible values they may have. First - Primary Quantum number ( $n$ ) = size of electron cloud  $n = 1$  up to  $\infty$ ....in reality  $n = 1 - 7$  Second - Azimuthal or Angular Momentum Quantum number ( $l$ ) = shape of electron cloud;  $l = 0$  up to  $(n-1)$ ....

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2a.  $1s^2 2s^2 2p^6 3s^2 3p^5$  2 2 6 2b.  $1s^2 2s^2 2p^6 3s^2 3p^6$  2 2 6 10c.

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