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Explain your answer. The forward reaction is favored. The weaker acid and weaker base are produced in the forward reaction. HCO_3^- competes more strongly with H_2O to donate a proton, and OH^- competes more strongly with CO_3^{2-} to acquire a proton, causing the forward reaction to be favored. MODERN CHEMISTRY ACIDS AND BASES 121

14 Acids and Bases - David Brearley High School

Sample Study Sheet 4.2: Abbreviated Electron Configurations To get a review of the most important topics in the chapter, fill in the blanks in the Key Ideas section. Work all of the selected problems at the end of the chapter, and check your answers with the solutions provided in this chapter of the study guide. Ask for help if you need it.

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Chapter 4 Modern Atomic Theory - An Introduction to Chemistry

to answer the following: Mass of the empty crucible 4.00 g Mass of the crucible plus hydrate sample 4.50 g Mass of the system after heating 4.32 g Mass of the system after a second heating 4.32 g 36% a. Determine the mass percentage of water in the original sample. MODERN CHEMISTRY CHEMICAL FORMULAS AND CHEMICAL COMPOUNDS59

7 Chemical Formulas and Chemical Compounds

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Teacher's ...

4 Study Guide for An Introduction to Chemistry Memorize the following. (Be sure to check with your instructor to determine how much you are expected to know of the following.) SI base units (Table 1.1) The table below contains the four base units that you should know now.

Chapter 1 An Introduction to Chemistry

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT
ANSWER Answer the following questions in the space provided.

1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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