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reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

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Chapter 12 Stoichiometry 127 SECTION  
12.1 THE ARITHMETIC OF EQUATIONS  
(pages 353–358) This section explains  
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Chapter 3 Stoichiometry: Calculations



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### Answers

with Chemical Formulas and Equations

3.1 Multiple -Choice Questions 1) When the following equation is balanced, the coefficients are \_\_\_\_\_.  $C_8H_{18} + O_2 \rightarrow CO_2 + H_2O$  A) 2, 3, 4, 4 B) 1, 4, 8, 9 C) 2, 12, 8, 9 D) 4, 4, 32, 36 E) 2, 25, 16, 18  
Answer: E Diff: 2 Page Ref: Sec. 3.1

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12.3 Limiting Reagent and Percent Yield  
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